

Structural and electronic properties of europium-doped SnO₂: first principle insights

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Abstract

The CRYSTAL17 code [1] was used to apply the density functional theory (DFT) to rutile tetragonal tin oxide (SnO₂) with space group P_{42}/mnm . First, the crystalline structure of SnO₂ was modeled in a unit cell, which contains 2 atoms of tin and 4 atoms of oxygen. We employed different bases using the PBE functional to have a greater agreement with the lattice constant experimental. Second, in order to investigate the impacts of the Eu³⁺ impurity, the SnO₂ structure, with an oxygen vacancy (V_O) and two europium (Eu) atoms, was meticulously modeled in a 3×3×2 supercell (Eu₂Sn₃₄O₇₁). The total energies of multiple Eu₂Sn₃₄O₇₁ supercell configurations were calculated and compared. It was discovered that when two Eu atoms are bonded through an oxygen atom and the V_O is bonded to an Eu atom, the total energy is at its lowest. To account for the six unpaired *f* electrons in the Eu³⁺ cations, all Eu-doped supercell simulations were spin-polarized. The electronic properties such as the band structure, density of state (DOS), bandgap, and effective mass were predicted using HSE06 functional. The band structure and DOS predicts that SnO₂ and Eu₂Sn₃₄O₇₁ supercells exhibit insulator character. Furthermore, the Eu₂Sn₃₄O₇₁ supercell permits intraband *f-f* transitions that are otherwise prohibited in atomic Eu. This result is consistent with photoluminescence (PL) found for Eu-doped SnO₂ (ETO) films and nanoparticles in the visible red range. The characteristics of the ETO compound suggest that it could be a suitable phosphor material for reducing thermalization losses in solar cells [2].

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References

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